

3,5-Dimethoxyacetanilide

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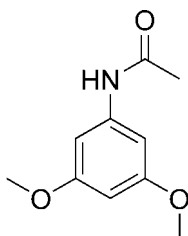
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.042; wR factor = 0.108; data-to-parameter ratio = 13.6.

The title compound, $\text{C}_{10}\text{H}_{13}\text{NO}_3$, crystallizes with two molecules in the asymmetric unit. Molecules related by an a -axis translation are stacked over each other, bound by π - π interactions with a perpendicular distance of 3.455 Å. Molecules in adjacent stacks are linked to each other through $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For related literature, see: Allen *et al.* (1987); Hall *et al.* (1992); Hunter & Sanders (1990); Traxler *et al.* (1999).



Experimental

Crystal data

$\text{C}_{10}\text{H}_{13}\text{NO}_3$	$V = 4003.4$ (6) Å ³
$M_r = 195.21$	$Z = 16$
Orthorhombic, $Pbca$	Mo $K\alpha$ radiation
$a = 7.1956$ (6) Å	$\mu = 0.10$ mm ⁻¹
$b = 16.8320$ (14) Å	$T = 293$ (2) K
$c = 33.054$ (3) Å	$0.30 \times 0.20 \times 0.20$ mm

Data collection

Bruker SMART APEXII diffractometer	19389 measured reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	3548 independent reflections
$T_{\min} = 0.972$, $T_{\max} = 0.981$	2808 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.033$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.042$	260 parameters
$wR(F^2) = 0.108$	H-atom parameters constrained
$S = 1.03$	$\Delta\rho_{\text{max}} = 0.25$ e Å ⁻³
3548 reflections	$\Delta\rho_{\text{min}} = -0.14$ e Å ⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{O6}^i$	0.86	2.03	2.8857 (18)	171
$\text{N2}-\text{H2A}\cdots\text{O3}^{ii}$	0.86	2.17	2.9799 (19)	158

Symmetry codes: (i) $-x + \frac{3}{2}, y + \frac{1}{2}, z$; (ii) $x - 1, y, z$.

Data collection: *APEX2* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Sheldrick, 2000); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: OM2120).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Bruker (2000). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Hall, R. J., Marchant, J., Oliveira-Campos, A. M. F., Queiroz, M.-J. R. P. & Shannon, P. V. R. (1992). *J. Chem. Soc. Perkin Trans. 1*, pp. 3439–3450.
- Hunter, C. A. & Sanders, J. K. M. (1990). *J. Am. Chem. Soc.* **112**, 5525–5534.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Sheldrick, G. M. (2000). *SHELXTL*. Version 6.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Traxler, P., Green, J., Mett, H., Séquin, U. & Furet, P. (1999). *J. Med. Chem.* **42**, 1018–1026.

supplementary materials

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3,5-Dimethoxyacetanilide

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Comment

The title compound, (I), is an important intermediate used to synthesize a variety of pharmaceuticals, such as epidermal growth factor receptor (EGFR) tyrosine kinase inhibitors (Traxler *et al.*, 1999). In our recent research for the synthesis of potential PDE5 inhibitors, 3,5-dimethoxyanilide (I) was synthesized as one of the structural units. We report here the crystal structure of the title compound, which crystallizes with two molecules in the asymmetric unit.

A view of the molecular structure of (I) is depicted in Fig. 1. In the two distinct molecules, the *N*-acetyl group is oriented at angles of 3.10 (2)° and 20.00 (3)° to the aromatic plane. All bond lengths and angles are normal (Allen *et al.*, 1987). Molecules related by an *a*-axis translation are stacked over each other and the perpendicular distance between the stacking planes is 3.455 Å (Fig. 2). It is, therefore, inferred that the stacked molecules are bound to each other by π - π interactions (Hunter & Sanders, 1990). The stacked columns are linked together *via* an intermolecular hydrogen bond in which the amide H act as a donor to O atom (Fig. 2 and Table 1).

Experimental

The title compound was prepared by acetylation of 3,5-dimethoxyaniline according to the method described by Hall *et al.*, (1992). Single crystals suitable for X-ray analysis (m.p. 426 K) were obtained by slow evaporation of a dichloromethane solution at 298 K.

Refinement

All H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C—H = 0.93 - 0.96 Å, N—H = 0.86 Å, and $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ or (N).

Figures

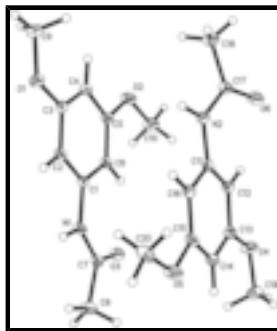


Fig. 1. View of the molecule of (I) showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

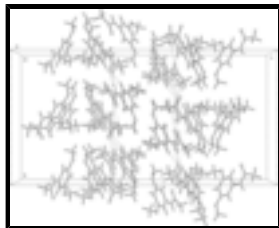


Fig. 2. The crystal packing of (I), viewed along the $\langle i \rangle a \langle i \rangle$ axis. Hydrogen bonds are shown as dashed lines.

3,5-Dimethoxyacetanilide

Crystal data

$C_{10}H_{13}NO_3$

$M_r = 195.21$

Orthorhombic, *Pbca*

$a = 7.1956 (6) \text{ \AA}$

$b = 16.8320 (14) \text{ \AA}$

$c = 33.054 (3) \text{ \AA}$

$V = 4003.4 (6) \text{ \AA}^3$

$Z = 16$

$F_{000} = 1664$

$D_x = 1.296 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

$\lambda = 0.71073 \text{ \AA}$

Cell parameters from 4958 reflections

$\theta = 2.5\text{--}24.7^\circ$

$\mu = 0.10 \text{ mm}^{-1}$

$T = 293 (2) \text{ K}$

Block, colourless

$0.30 \times 0.20 \times 0.20 \text{ mm}$

Data collection

Bruker SMART APEXII
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 298(2) \text{ K}$

φ and ω scans

Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)

$T_{\min} = 0.972$, $T_{\max} = 0.981$

19389 measured reflections

3548 independent reflections

2808 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.033$

$\theta_{\max} = 25.1^\circ$

$\theta_{\min} = 2.5^\circ$

$h = -8 \rightarrow 8$

$k = -19 \rightarrow 20$

$l = -30 \rightarrow 39$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.042$

$wR(F^2) = 0.108$

$S = 1.03$

3548 reflections

260 parameters

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.047P)^2 + 1.2714P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.25 \text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -0.14 \text{ e \AA}^{-3}$

Extinction correction: SHELXL97,
 $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.0015 (4)

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.9040 (2)	0.66216 (10)	0.91166 (5)	0.0360 (4)
C2	0.7495 (2)	0.70799 (10)	0.92004 (5)	0.0416 (4)
H2	0.7399	0.7590	0.9095	0.050*
C3	0.6087 (2)	0.67783 (11)	0.94428 (5)	0.0430 (4)
C4	0.6212 (3)	0.60184 (11)	0.96012 (5)	0.0464 (5)
H4	0.5277	0.5817	0.9766	0.056*
C5	0.7761 (3)	0.55664 (10)	0.95083 (5)	0.0428 (4)
C6	0.9191 (2)	0.58522 (10)	0.92700 (5)	0.0404 (4)
H6	1.0226	0.5540	0.9214	0.048*
C7	1.2042 (2)	0.66862 (10)	0.87336 (5)	0.0410 (4)
C8	1.3073 (3)	0.72307 (12)	0.84542 (7)	0.0596 (6)
H8A	1.4291	0.7331	0.8561	0.089*
H8B	1.2409	0.7723	0.8430	0.089*
H8C	1.3181	0.6987	0.8193	0.089*
C9	0.3195 (3)	0.70352 (14)	0.97704 (7)	0.0659 (6)
H9A	0.2630	0.6562	0.9665	0.099*
H9B	0.2275	0.7446	0.9792	0.099*
H9C	0.3705	0.6926	1.0033	0.099*
C10	0.9237 (3)	0.42958 (11)	0.95603 (7)	0.0647 (6)
H10A	0.9281	0.4240	0.9271	0.097*
H10B	0.9047	0.3784	0.9682	0.097*
H10C	1.0388	0.4516	0.9655	0.097*
C11	0.7292 (2)	0.50583 (10)	0.83936 (5)	0.0371 (4)
C12	0.8709 (2)	0.45172 (10)	0.83355 (5)	0.0412 (4)
H12	0.8650	0.4016	0.8454	0.049*
C13	1.0227 (2)	0.47288 (10)	0.80982 (5)	0.0420 (4)
C14	1.0337 (3)	0.54668 (11)	0.79191 (5)	0.0460 (4)
H14	1.1359	0.5604	0.7761	0.055*

supplementary materials

C15	0.8897 (3)	0.60002 (10)	0.79786 (5)	0.0454 (5)
C16	0.7373 (3)	0.58076 (10)	0.82121 (5)	0.0426 (4)
H16	0.6414	0.6171	0.8248	0.051*
C17	0.5121 (2)	0.41904 (10)	0.87722 (5)	0.0424 (4)
C18	0.3505 (3)	0.42184 (12)	0.90562 (7)	0.0607 (6)
H18A	0.2966	0.3698	0.9078	0.091*
H18B	0.2591	0.4583	0.8955	0.091*
H18C	0.3919	0.4392	0.9318	0.091*
C19	1.3080 (3)	0.43027 (14)	0.78001 (6)	0.0651 (6)
H19A	1.2630	0.4393	0.7530	0.098*
H19B	1.3903	0.3854	0.7800	0.098*
H19C	1.3736	0.4765	0.7893	0.098*
C20	0.7776 (3)	0.73162 (12)	0.78511 (7)	0.0663 (6)
H20A	0.6589	0.7135	0.7756	0.099*
H20B	0.8127	0.7787	0.7706	0.099*
H20C	0.7697	0.7434	0.8135	0.099*
N1	1.03993 (19)	0.69747 (8)	0.88680 (4)	0.0405 (4)
H1	1.0144	0.7449	0.8789	0.049*
N2	0.5737 (2)	0.49023 (8)	0.86409 (4)	0.0418 (4)
H2A	0.5105	0.5310	0.8717	0.050*
O1	0.46382 (18)	0.72883 (8)	0.95061 (4)	0.0596 (4)
O2	0.77427 (19)	0.48116 (8)	0.96687 (4)	0.0603 (4)
O3	1.26690 (17)	0.60353 (7)	0.88250 (4)	0.0516 (4)
O4	1.15542 (19)	0.41493 (8)	0.80618 (4)	0.0606 (4)
O5	0.9123 (2)	0.67166 (8)	0.77872 (5)	0.0684 (4)
O6	0.58211 (19)	0.35588 (7)	0.86658 (4)	0.0593 (4)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0339 (9)	0.0350 (9)	0.0390 (9)	-0.0056 (7)	-0.0013 (7)	0.0005 (7)
C2	0.0427 (10)	0.0338 (9)	0.0483 (10)	-0.0018 (8)	0.0016 (8)	0.0004 (8)
C3	0.0389 (10)	0.0420 (10)	0.0481 (10)	-0.0026 (8)	0.0049 (8)	-0.0096 (8)
C4	0.0449 (11)	0.0473 (11)	0.0472 (11)	-0.0106 (9)	0.0100 (9)	-0.0012 (9)
C5	0.0464 (10)	0.0379 (10)	0.0442 (10)	-0.0077 (8)	0.0019 (9)	0.0046 (8)
C6	0.0393 (9)	0.0346 (9)	0.0472 (10)	-0.0029 (8)	0.0015 (8)	0.0032 (8)
C7	0.0388 (9)	0.0355 (10)	0.0488 (10)	-0.0020 (8)	0.0023 (8)	0.0045 (8)
C8	0.0496 (12)	0.0519 (12)	0.0772 (14)	0.0034 (10)	0.0167 (11)	0.0182 (10)
C9	0.0507 (12)	0.0783 (16)	0.0687 (14)	-0.0001 (11)	0.0212 (11)	-0.0086 (12)
C10	0.0717 (14)	0.0412 (11)	0.0811 (15)	0.0054 (10)	0.0221 (12)	0.0182 (10)
C11	0.0434 (10)	0.0345 (9)	0.0334 (9)	-0.0049 (8)	0.0004 (8)	-0.0004 (7)
C12	0.0476 (10)	0.0356 (9)	0.0402 (10)	-0.0011 (8)	0.0036 (8)	0.0060 (8)
C13	0.0433 (10)	0.0430 (10)	0.0395 (10)	0.0013 (8)	0.0008 (8)	0.0013 (8)
C14	0.0432 (10)	0.0491 (11)	0.0458 (10)	-0.0093 (9)	0.0047 (8)	0.0052 (9)
C15	0.0540 (11)	0.0349 (10)	0.0473 (10)	-0.0089 (9)	-0.0012 (9)	0.0072 (8)
C16	0.0475 (10)	0.0328 (9)	0.0474 (10)	-0.0007 (8)	0.0019 (9)	0.0020 (8)
C17	0.0418 (10)	0.0371 (10)	0.0484 (11)	-0.0006 (8)	0.0018 (9)	0.0091 (8)
C18	0.0609 (13)	0.0515 (12)	0.0698 (14)	0.0009 (10)	0.0218 (11)	0.0146 (10)

C19	0.0476 (12)	0.0824 (16)	0.0652 (14)	0.0079 (11)	0.0153 (11)	0.0046 (12)
C20	0.0742 (15)	0.0407 (11)	0.0838 (16)	-0.0050 (11)	-0.0032 (13)	0.0169 (11)
N1	0.0377 (8)	0.0306 (7)	0.0533 (9)	0.0007 (6)	0.0051 (7)	0.0097 (6)
N2	0.0471 (9)	0.0312 (8)	0.0473 (9)	0.0012 (7)	0.0114 (7)	0.0023 (6)
O1	0.0494 (8)	0.0505 (8)	0.0788 (10)	0.0038 (6)	0.0238 (7)	-0.0031 (7)
O2	0.0619 (9)	0.0446 (8)	0.0742 (9)	-0.0014 (7)	0.0207 (8)	0.0197 (7)
O3	0.0492 (8)	0.0363 (7)	0.0694 (9)	0.0076 (6)	0.0107 (7)	0.0104 (6)
O4	0.0530 (8)	0.0597 (9)	0.0691 (9)	0.0127 (7)	0.0207 (7)	0.0153 (7)
O5	0.0704 (10)	0.0442 (8)	0.0907 (11)	-0.0032 (7)	0.0199 (8)	0.0249 (7)
O6	0.0538 (8)	0.0321 (7)	0.0920 (11)	0.0036 (6)	0.0194 (8)	0.0095 (7)

Geometric parameters (Å, °)

C1—C2	1.381 (2)	C11—C16	1.398 (2)
C1—C6	1.395 (2)	C11—N2	1.410 (2)
C1—N1	1.409 (2)	C12—C13	1.391 (2)
C2—C3	1.388 (2)	C12—H12	0.9300
C2—H2	0.9300	C13—O4	1.371 (2)
C3—O1	1.366 (2)	C13—C14	1.378 (2)
C3—C4	1.385 (3)	C14—C15	1.385 (3)
C4—C5	1.384 (3)	C14—H14	0.9300
C4—H4	0.9300	C15—O5	1.371 (2)
C5—O2	1.377 (2)	C15—C16	1.379 (2)
C5—C6	1.382 (2)	C16—H16	0.9300
C6—H6	0.9300	C17—O6	1.228 (2)
C7—O3	1.223 (2)	C17—N2	1.349 (2)
C7—N1	1.353 (2)	C17—C18	1.495 (3)
C7—C8	1.498 (2)	C18—H18A	0.9600
C8—H8A	0.9600	C18—H18B	0.9600
C8—H8B	0.9600	C18—H18C	0.9600
C8—H8C	0.9600	C19—O4	1.421 (2)
C9—O1	1.423 (2)	C19—H19A	0.9600
C9—H9A	0.9600	C19—H19B	0.9600
C9—H9B	0.9600	C19—H19C	0.9600
C9—H9C	0.9600	C20—O5	1.415 (3)
C10—O2	1.428 (2)	C20—H20A	0.9600
C10—H10A	0.9600	C20—H20B	0.9600
C10—H10B	0.9600	C20—H20C	0.9600
C10—H10C	0.9600	N1—H1	0.8600
C11—C12	1.381 (2)	N2—H2A	0.8600
C2—C1—C6	120.54 (16)	C13—C12—H12	120.3
C2—C1—N1	116.11 (14)	O4—C13—C14	124.29 (16)
C6—C1—N1	123.34 (15)	O4—C13—C12	114.49 (15)
C1—C2—C3	119.96 (16)	C14—C13—C12	121.23 (17)
C1—C2—H2	120.0	C13—C14—C15	118.67 (17)
C3—C2—H2	120.0	C13—C14—H14	120.7
O1—C3—C4	124.89 (16)	C15—C14—H14	120.7
O1—C3—C2	114.56 (16)	O5—C15—C16	124.01 (17)
C4—C3—C2	120.56 (17)	O5—C15—C14	114.53 (16)

supplementary materials

C5—C4—C3	118.43 (16)	C16—C15—C14	121.46 (16)
C5—C4—H4	120.8	C15—C16—C11	119.05 (17)
C3—C4—H4	120.8	C15—C16—H16	120.5
O2—C5—C6	123.21 (16)	C11—C16—H16	120.5
O2—C5—C4	114.46 (16)	O6—C17—N2	122.81 (16)
C6—C5—C4	122.32 (16)	O6—C17—C18	121.76 (16)
C5—C6—C1	118.18 (16)	N2—C17—C18	115.43 (16)
C5—C6—H6	120.9	C17—C18—H18A	109.5
C1—C6—H6	120.9	C17—C18—H18B	109.5
O3—C7—N1	124.26 (16)	H18A—C18—H18B	109.5
O3—C7—C8	121.16 (16)	C17—C18—H18C	109.5
N1—C7—C8	114.57 (15)	H18A—C18—H18C	109.5
C7—C8—H8A	109.5	H18B—C18—H18C	109.5
C7—C8—H8B	109.5	O4—C19—H19A	109.5
H8A—C8—H8B	109.5	O4—C19—H19B	109.5
C7—C8—H8C	109.5	H19A—C19—H19B	109.5
H8A—C8—H8C	109.5	O4—C19—H19C	109.5
H8B—C8—H8C	109.5	H19A—C19—H19C	109.5
O1—C9—H9A	109.5	H19B—C19—H19C	109.5
O1—C9—H9B	109.5	O5—C20—H20A	109.5
H9A—C9—H9B	109.5	O5—C20—H20B	109.5
O1—C9—H9C	109.5	H20A—C20—H20B	109.5
H9A—C9—H9C	109.5	O5—C20—H20C	109.5
H9B—C9—H9C	109.5	H20A—C20—H20C	109.5
O2—C10—H10A	109.5	H20B—C20—H20C	109.5
O2—C10—H10B	109.5	C7—N1—C1	130.28 (14)
H10A—C10—H10B	109.5	C7—N1—H1	114.9
O2—C10—H10C	109.5	C1—N1—H1	114.9
H10A—C10—H10C	109.5	C17—N2—C11	127.76 (15)
H10B—C10—H10C	109.5	C17—N2—H2A	116.1
C12—C11—C16	120.29 (16)	C11—N2—H2A	116.1
C12—C11—N2	122.95 (15)	C3—O1—C9	117.57 (16)
C16—C11—N2	116.74 (15)	C5—O2—C10	117.22 (14)
C11—C12—C13	119.30 (16)	C13—O4—C19	117.55 (15)
C11—C12—H12	120.3	C15—O5—C20	118.46 (16)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots O6 ⁱ	0.86	2.03	2.8857 (18)	171
N2—H2A \cdots O3 ⁱⁱ	0.86	2.17	2.9799 (19)	158

Symmetry codes: (i) $-x+3/2, y+1/2, z$; (ii) $x-1, y, z$.

Fig. 1

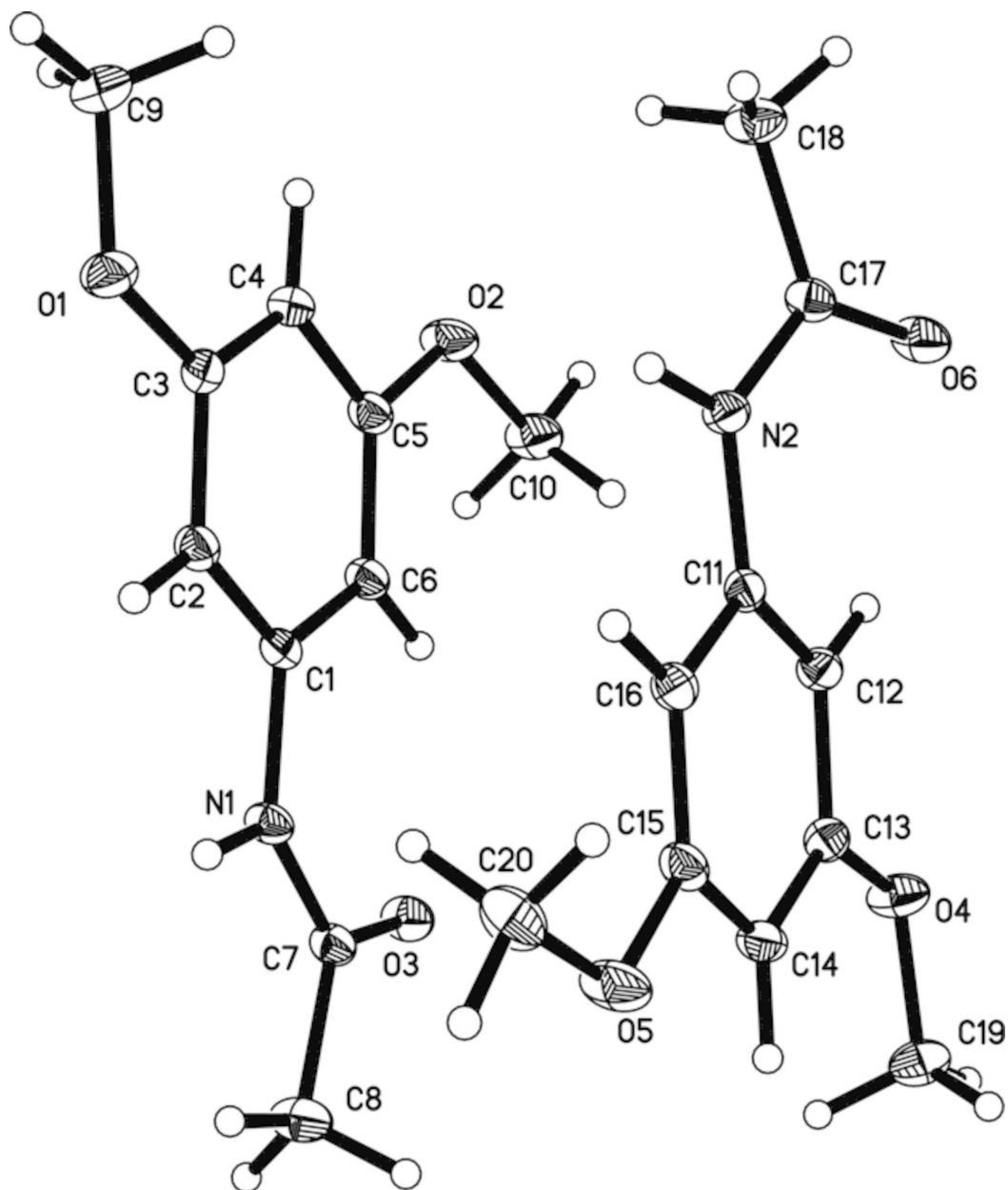


Fig. 2

